Polytypism in Sodium 2-Oxocaproate Crystals*

By L. M. Pant

National Chemical Laboratory, Poona, India

(Received 30 December 1966 and in revised form 22 September 1967)

Sodium 2-oxocaproate crystals contain ordered, disordered and polytypic regions in them. The type of disorder and the structure of the polytype are described.

Crystals of sodium 2-oxocaproate,

$CH_3(CH_2)_3COCOONa$,

give a pattern of sharp and diffuse X-ray reflexions. It is seen from zero and higher layer Weissenberg photographs that the reflexions with h+k+l even and k+l even are sharp while those with h+k+l even and k+l odd, although sharp, appear above a background of diffuse maxima. Further, reflexions corresponding to an *a* axial length twice that of the normal cell appear on either side of the reflexions with h+k+l even and k+l odd; with respect to the double cell, these reflexions have the indices Hkl with H odd. Figs. from the equivalent points of A and B that if the B layer is displaced by $\frac{1}{2}(\mathbf{b}+\mathbf{c})$, the unit cell becomes AA instead of AB. For such disorder, it can be shown (Pant, 1964) that as observed, reflexions with h+k+l even and k+l even are sharp while those with h+k+l even (*i.e.* H/2+k+l even) and k+l odd are diffuse; the sharp reflexions appearing above the background of diffuse maxima (*e.g.* reflexions marked 30.1.0 and 30.3.0 in Fig. 2(*a*) and (*b*) respectively) come from the ordered region in the crystal.

The structure of the polytype is, as shown later, *AABBAABB*... The equivalent points for the unit cell of the polytype are as follows:

1 and 2 show reflexions on zero layer Weissenberg photographs taken with **c** as oscillation axis; the indices Hk0 shown are with respect to the double cell, *i.e.* H=2h. These results suggest that the crystals contain ordered, disordered and polytypic regions in them. In this communication, the type of disorder and the structure of the polytype are described.

The data already reported for the basic structure (Jain, Tavale & Biswas, 1966) are: a=39.46, b=6.11, c=5.94 Å; space group *Iba2*. This space group has the following eight equivalent positions:

Each equivalent position corresponds to one molecule; the four molecules marked A form the repeat unit of one layer of molecules while the other four marked B form the repeat unit of the neighbouring layer. The disorder in the crystal consists in the appearance of any layer as displaced from its expected position in the ordered structure by $\frac{1}{2}(\mathbf{b}+\mathbf{c})$. It is clear In the above, X corresponds to the unit cell of the polytype, therefore x=2X, since x corresponds to the unit cell of the basic structure. It is obvious from the equivalent points that the space group of the polytype is also *Iba2*, and the asymmetric unit consists of two molecules, one represented by (X, y, z) and the other by $(\frac{1}{4}+X, y, z)$.

From the structure factor formulae obtained with the use of the equivalent points written above, it may be shown for reflexions with k+l even that as observed, reflexions from the polytype would be present for H=4n only, *i.e.* for h=2n. These reflexions coincide with and appear with the same absolute intensity as the reflexions from the normal crystal. Similarly, it may be shown for reflexions with k+l odd that only those with H odd would be present. With respect to the unit cell of the basic structure, these reflexions have the indices $h \pm \frac{1}{2}$, k, l, and they appear on either side of the reflexions from the basic structure (Fig. 2). The structure factor formulae for the polytype for reflexions of the type Hk0 are as follows:

$$F(Hk0) = 8 \cos 2\pi ky \{ \cos 2\pi HX - \sin 2\pi HX \}$$

for k odd and $H = 4n + 1$, and
 $8 \cos 2\pi ky \{ \cos 2\pi HX + \sin 2\pi HX \}$
for k odd and $H = 4n - 1$.

For k odd and H even, F(Hk0) = 0.

^{*} Communication No. 1032 from the National Chemical Laboratory, Poona, India.

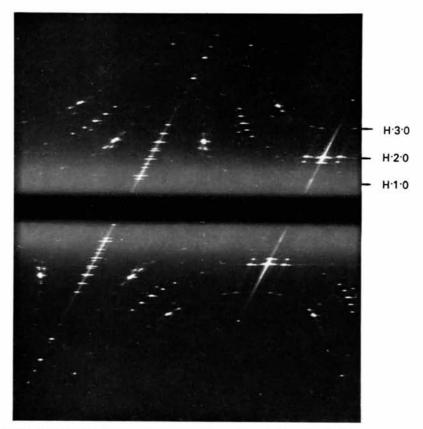


Fig. 1. Zero layer Weissenberg photograph. Oscillation axis c. Cu radiation, 30kV, 17mA, exposure 16hr.

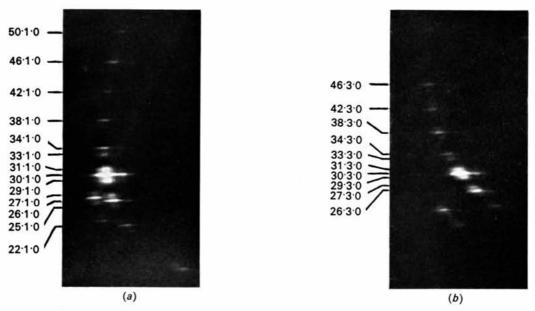


Fig. 2. Enlargements of parts of Fig. 1, showing (a) H10 reflexions, (b) H30 reflexions.

Table 1. Observed and calculated structure factors

H k	0	$ F_o $	Fc	H	k	0	$ F_o $	F_{c}
23 1	0	28	38	23	3	0	23	38
25 1	0	47	49	25	3	0	42	$\frac{38}{41}$
27 1 29 1	0	98	92	27	3	0	93	89 144
29 1	0	158	135	29	3	0	122	144
31 1 33 1	0	157 87	131 75	31	3	0	146	$ \begin{array}{r} 142 \\ \overline{72} \\ 12 \\ \overline{53} \\ \overline{50} \\ 40 \\ 44 \\ \overline{50} \\ \overline{43} \\ \end{array} $
33 1	0	87	75	33	3	0	79	72
35 1	0	< 26	10	35	3	· 0	< 28	12
37 1	0	26	25	37	3	0	49	53
39 1	0 0	39	29	39	3	0	49 51	50
41 1	0	< 28	29 37	41	3	0	31	40
43 1	0	40	37	43	3	0	31	44
45 1	0	42	44	45	3	0	54	30
47. 1	0	43	40	47	3	0	45	<u>43</u>

With the help of these formulae, some structure factors have been computed from the unpublished data of Jain, Tavale & Biswas (see Appendix). These are compared in Table 1 with the observed structure factors. F_o were obtained from visually estimated intensities (after being corrected for the Lp factor); the scaling was done with the help of F_c . The agreement between F_o and F_c shows that the postulated structure of the polytype is correct.

The only other possibility of having a polytype from A and B layers with an a axial length which is twice that of the normal cell, is to have a structure with three layers of one type and the fourth layer of the other type, say AAAB; the other alternatives like AABA, BBBA, etc. may be obtained from AAAB by suitably changing the origin. However, this possibility is ruled out since for this structure, F(Hk0) for k odd and H=4n are, in general, present whereas such reflexions are actually found to be systematically absent (Fig. 2). As a further check, the intensity of the 28,1,0 reflexion; however, in Fig.2(a), the 27,1,0 reflexion can be clearly seen whereas 28,1,0 is absent.

My thanks are due to Mr S.C. Jain, Dr S.S. Tavale and Dr A.B. Biswas for permitting me to use their unpublished data. I am also grateful to Dr T.D.Sakore for help in the experimental work and to Mr S.C.Jain for doing the computations in the electronic computer.

References

JAIN, S. C., TAVALE, S. S. & BISWAS, A. B. (1966). Acta Cryst. 21, 445.

PANT, L. M. (1964). Acta Cryst. 17, 219.

APPENDIX

BY S.C. JAIN, S.S. TAVALE & A.B. BISWAS

Atomic and thermal parameters in the basic structure

	x	У	Z	В
Na	0.2215	0.2195	-0.0064	3·1 Å2
O(1)	0.2230	0.1563	0.3830	2.4
O(2)	0.1639	0.1592	0.1692	3.7
O(3)	0.2015	0.3802	0.6519	3.0
C(1)	0.2016	0.2492	0.4873	3.0
C(2)	0.1652	0.2235	0.3516	3.3
C(3)	0.1343	0.3083	0.4888	3.7
C(4)	0.1022	0.2066	0.3712	4 ·0
C(5)	0.0709	0.3156	0.5206	4.5
C(6)	0.0375	0.1944	0.4091	4.8